# OLIGOMERIC FLAVANOIDS. PART 14'. PROGUIBOURTINIDINS BASED ON (-)-FISETINIDOL AND (+)-EPIFISETINIDOL UNITS.

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Abstract  $-$  The rare series of proguibourtinidins  $(3,4',7$ -trihydroxy functionality) based on the fisetinidol (3,3',4',7-tetrahydroxy functionality) 5-deoxy flavan-3-01 unit is extended by identification of  $(+)$ -guibourtinidol- $(4\beta, 6)$ - $(-)$ -fisetinidol and the  $(+)$ -guibourtini $dol-(4\alpha,6)$  and  $(4\beta,6)-(+)$ -epifisetinidols. These metabolites are formed in low yields via acid-catalyzed coupling of (+)-guibourtacacidin and  $(-)$ -fisetinidol, the high activation energy of generating a 4-C benzylic carbocation being attributable to the poor electron donating properties of the phenol B-ring of the flavan-3,4-diol. Stereo-electronic principles operating in the **in vitro** process presumably also control their formation in Nature.

Pro- and leucoguibourtinidins with their 4' ,7-dihydroxy phenolic functionality represent a relatively rare group of condensed tannins which, while occurring as minor components in Australian Acacia spp.<sup>1</sup>, predominate in the Southern African species Guibourtia coleoaperad (large false mopane) *2\*3, Julb ernardia globiflora* (munondo)', and Acacia Iuederiltii (bastard umbrella thorn)<sup>5,6</sup>. Known analogues from the last three sources are invariably based on 5-oxygenated 'lower' flavan-3-01 units, i.e. (t)-catechin, (-)-epicatechin, and (t)-afzelechin. Our recent demonstration of the occurrence of (t)-guibourtinidol-(-) fisetinidols in the heartwood of *Colophospermum mopane* (mopane)<sup>7,8</sup>, reputed for its exceptionally high concentrations<sup>7</sup> of the 5-deoxy flavan-3-ols (-)-fisetinidol<sup>b</sup> and (+)epifisetinidol<sup>b</sup>, indicated that these 5-deoxy analogues may feasibly act as nucleophiles in the biosynthetic pathway leading to this class of proguibourtinidins. A systematic approach towards the chemistry and extension of this rare series of oligoflavanoids thus became objectives in our continuing investigation of the diverse metabolic pool of the mopane.

<sup>=</sup>Part 13. J.C.S. Malan, J.A. Steenkamp, D.A. Young, and D. Ferreira, Tetrahedron, 1989, paper AM 9295.

 $b(-)$ -Fisetinidol is  $(2\ell,3S)$ -2,3- $irans-flavan-3,3',4',7-tetraol and (+)-epifisetinidol the$  $(25,35)-2,3-cis$  C-2 epimer.

### RESULTS AND DISCUSSION

The  $(+)$ -guibourtinidol- $(4\alpha, 6)$ - $(-)$ -fisetinidol  $1^7$ ,  $(+)$ -guibourtinidol- $(4\alpha, 8)$ - $(-)$ -fisetinidol  $9^8$  and (+)-guibourtinidol-(4 $\sigma$ ,6')-(-)-fisetinidol  $10^8$  are accompanied in the heartwood of  $C$ . mopane by the  $(+)$ -guibourtinidol- $(4\beta, 6)$ - $(-)$ -fisetinidol  $2$  and the  $(+)$ -guibourtinidol- $(4a,6)$  and  $(4\beta,6)$ - $(+)$ -epifisetinidols 5 and 7. Owing to the complexity of the phenolic mixture these novel netabolites were identified as pentamethyl ether diacetates,  $e.g. 4$ , the additional chromatographic stages offered by such an approach being a prerequisite for compound purity.



Comparison of the <sup>1</sup>H NMR data (Table) of the pentamethyl ether diacetate  $4$  of the  $(+)$ guibourtinidol- $(4\beta, 6)$ - $(-)$ -fisetinidol 2 with those of the  $(4\sigma, 6)$ -analogue 2 reveals their close structural resemblance. The relative  $2,3-trans-3,4-cis$  (C-ring):  $2,3-trans$  (F) configuration is evident from the coupling constants of heterocyclic protons IJi,s 7.5, J3,4

**5.0 Hz;**  $J_2, g(F)$  7.5 Hz]. Correlation of the  $AA'BB'$  system in the aromatic region  $(\delta7.27,$ **6.85, J9.0** Hz) with the 2-H doublet (65.14) of the heterocyclic AMX system and of 4-H (64.79) of the latter system with 5-H(A) (66.77, d, 58.5 Hz) of the highfield aromatic **ABX**  system **via** decoupling experiments, defines the constitution of the (+)-guibourtinidol unit. Substitution at 6-C of the 'lower'  $2,3-trans$  fisetinidol moiety is confirmed by the characteristic appearance of the D-ring protons as singlets<sup>7</sup> [ $\delta$ 6.55 (br.), 6.46; 5- and 8-H(D) respectively]. NOE experiments not only confirmed these allocations but also facilitated assignment of all methoxy resonances. The same protocol of decoupling experiments using the 2-(C- and F) and 4-H(C) resonances as reference signals, supported by appropriate NOE data, also enables structural definition of the pentamethyl ether diacetates  $\underline{6}$  and  $\underline{8}$  of the novel (+)-guibourtinidol-(4 $\sigma$ ,6) and (4 $\beta$ ,6)-(+)-epifisetinidols  $\frac{1}{2}$  and  $\frac{1}{2}$  [J<sub>2</sub>,<sub>3</sub>(F) *ca.* 1.0 Hz for the **2,3-cis** DEF moieties of both 5 **and 81 (Table).** 

Table <sup>I</sup>H NMR peaks (p.p.m.) of (+)-guibourtinidol-fisetinidol derivatives 2, 4, 6, **and 8 at 300 MHz** in CDC13. Splitting patterns and J-values (Hz) **are** given **in parentheses.** 

Ring	H	2, 365 K	4, 296 K	6,345K	8, 296 K
$\mathbf{A}$	5 6	6.65(d, 8.5) $6.42$ (dd, $2.5, 8.5$ )	6.77(d,8.5) $6.46$ (dd, $2.5, 8.5$ )	6.67(d, 8.5) $6.43$ (dd, $2.5, 8.5$ )	6.80(d, 8.5) $6.47$ (dd, $2.5, 8.5$ )
	8	6.50(d, 2.5)	6.55(d, 2.5)	6.51(d, 2.5)	6.55(d, 2.5)
B	2/6	7.32(d, 9.0)	7.27(d, 9.0)	7.32(d, 9.0)	7.26(d, 9.0)
	3/5	6.84(d, 9.0)	6.85(d, 9.0)	6.84(d, 9.0)	6.85(d, 9.0)
$\mathbf{c}$	2 $\bf 3$ $\overline{\mathbf{4}}$	4.99(d, 9.0) $5.66$ (dd, $9.0, 9.0$ ) 4.54(d, 9.0)	5.14(d,7.5) 5.49(dd, 5.0, 7.5) 4.79(d, 5.0)	5.01(d, 9.0) 5.68(t, 9.0) 5.47(d, 9.0)	5.16(d, 7.5) 5.50(dd, 5.0, 7.5) 4.77(d, 5.0)
D	5 8	6.63(br.s) 6.46(s)	6.55(br.s) 6.46(s)	6.63(br.s) 6.49(s)	6.59(br.s) 6.51(s)
E	2 5 6	6.90(d, 2.0) 6.83(d,7.0) $6.89$ (dd, $2.0, 7.0$ )	6.88(d, 2.0) 6.82(d, 8.5) 6.91(dd, 2.0, 8.5)	7.01(d, 2.0) 6.84(d, 8.5) $6.94(d_1, 2.0, 8.5)$	7.01(d, 2.1) 6.85(d, 8.5) 6.94(dd, 2.1, 8.5)
$\mathbf{F}$	$\frac{2}{3}$ $4_{ex}$ $4_{eq}$	4.95(d,7.0) $5.27($ <b>m</b> ) $2.92$ (dd, $5.0, 16.0$ ) $2.69$ (dd, $7.0$ , $16.0$ )	4.97(d, 7.5) 5.28(n) $2.74$ (dd, $8.0, 16.0$ ) $2.95$ (dd, $5.5$ , $16.0$ )	5.03(br.s, ca 1.0) 5.37(m) 2.74(dd, 5.0, 17.5) $3.10$ (dd, 4.5, 17.5)	$5.03(br.s., c\leq 1.0)$ 5.38(m) $2.77$ (dd, $2.2$ , $17.5$ ) 3.15(dd, 4.5, 17.5)
	OMe	$3.74(7-D)$ , $3.76(7-$ $A), 3.79(4-B), 3.86$ $(3-E), 3.88(4-E)$ (each s)	$3.71(7-D)$ , $3.77(7-$ $A$ , 3.78(4-B), 3.84 $(3-E), 3.86(4-E)$ $\left($ each $s\right)$	$3.76(7-A/7-D)$ , $3.79$ $(4-B)$ , 3.86 $(4-E)$ , $3.87(3-E)(each s)$	$3.74(7-D)$ , $3.78(4-B)$ , $3.79(7-A), 3.88(4-E),$ $3.90(3-E)(each s)$
	<b>OAc</b>	$1.66, 1.88$ (each s)	1.76, 1.90 (each s)	1.66, 1.88 (each s)	1.76, 1.94 (each s)

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The absolute configurations, i.e.  $2l, 3S, 4S$  for  $2$  and  $2$ , and  $2l, 3S, 4l$  for  $5$ , of the proguibourtinidin moieties are evident' from **the** sign of the high-amplitude Cotton effects (positive **for 4** and 8, negative for 8) in the 225-240 na region of the CD spectra of their derivatives, and the known relative configurations as defined by coupling constants of the heterocyclic AMX protons. Confirmation of the  $(2\ell,3S)-(-)$ -fisetinidol (for 3) and  $(2S,3S)-$ (+)-epifisetinidol (for  $\underline{5}$  and  $\underline{7}$ ) 'terminal' units was obtained via their biomimetic synthesis.

Thus, condensation<sup>10</sup> of (+)-guibourtacacidin 11 [(2*l*, 3*5*, 4*l*)-2, 3-*trans*-3, 4-*trans*-flavan-3,4,4',7-tetraol]<sup>3</sup> and  $(-)$ -fisetinidol 12 in 0.1M HCl for 140h at 45<sup>0</sup>C affords a mixture com-



**(11)** R'= OH, Ra Ii  $(12)$  R'= H, R<sup>2</sup>= OH  $(13)$   $R1 = R1 = OM$ 

prising considerable quantities of  $(-)$ -fisetinidol and  $(+)$ -epifisetinidol (ca 4:1), and the proguibourtinidins 1, 3, 5, and 7 in 18.8% overall yield<sup>c</sup>. Their pentamethyl ether diacetates are identical to those of the natural products by comparison of IH NMR (Table) and **CD**  data.

Notable in the above conversion is the prolonged reaction time, conspicuously low yields, more severe conditions compared to those for the condensation of (+)-mollisacacidin 13 [3'-oxy analogue of (+)-guibourtacacidin 11] with (+)-catechin<sup>10</sup> (20<sup>0</sup>C, 2h, 45% yield) and (-)-fisetinidol<sup>11</sup> 12 (40<sup>o</sup>C, 24h, 36% yield), and the predominant formation of the  $(4\beta,6)$ -analogues 3 and 7 relative to that of the  $(4\sigma,6)$ -isomers 1 and 5  $(c\alpha$  4:1). Since the prevailing conditions do no permit interconversion of the  $(4\beta, 6)$ - and  $(4\alpha, 6)$ -analogues, the observed product ratio presumably reflects kinetic control<sup>12</sup> of coupling of the flavan- $3$ -ol moiety to the carbocationic intermediates  $16$  and  $18$  (compare, however, ref. 13). Whereas the enhanced rate of the reaction of  $(+)$ - $\mathbf{w}$ -llisacacidin 13 with  $(+)$ -catechin

CTr i- and tetrn-flavnnoid analogues (cf. **ref.** 11) are formed in proportions **which did** not merit further investigation,

compared to that with (-)-fisetinidol 12 is attributable to the superior nucleophilicity of the phloroglucinol-type A-ring of  $(+)$ -catechin vs the resorcinol-type A-ring of  $(-)$ fisetinidol, differences in the relative rates of condensation of  $(+)$ -mollisacacidin 13 and  $(+)$ -guibourtacacidin 11 with  $(-)$ -fisetinidol 12 are to be sought in the rates of formation of the 4-C carbocationic intermediates  $16$  and  $17$  derived from the flavan-3,4-diols. The significant difference may be rationalized on the assumption that the A-ring delocalized 4-carbocations which are generated under acidic conditions, are stabilized to varying degrees by delocalization of the benzylic charge over the B-ring. Although heterocyclic oxonium ions of types 14 and 15 were invoked $^{14}$  to explain the higher rate of condensation of  $(+)$ -leucorobinetinidin $[(2I,3S,4I)-2,3-trans-3,4-trans-flawan-3,3',4,4',5',7-hexa0!]$  with (+)-catechin compared to that of  $(+)$ -mollisacacidin 13, recent demonstration of the conformational mobility of the pyran heterocycle<sup>15-17</sup> indicates that the benzylic carbocations 16 and 17 may be additionally stabilized by charge donation from the B-ring via A-conformations 18 and 19 (Scheme). The more electron-rich pyrocatechol function in  $19$  is more effective than the mono-oxygenated moiety in J& thus leading to higher condensation rates for  $(+)$ -mollisacacidin 13.





 $(20)$ 

The relative drastic reaction conditions required for inducing formation of the 4-C carbocation  $16$  also leads to protonation of the  $(-)$ -fisetinidol heterocyclic oxygen hence initiating its conversion to the 2-C epimer, (+)-epifisetinidol, by recyclization via

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2-OH(A) and the Si-face at 2-C in a presumed quinone-methide intermediate 20. Replacement of (-)-fisetinidol by (t)-epifisetinidol as nucleophile in stereoselective coupling with the benzylic carbocation 16 may then explain the genesis of the  $(+)$ -guibourtinidol- $(4\sigma,6)$  and  $(4\beta, 6)-(+)$ -epifisetinidols  $\overline{2}$  and  $\overline{1}$ . Such epimerization at 2-C of the  $(-)$ -fisetinidol moiety may, however, also occur at the 'dimeric' level, e.g.  $1 \rightarrow 5$  and  $2 \rightarrow 7$ , by a similar mechanism.

A notable feature regarding the acid-catalyzed epimerizations is the apparent absence of a similar phenomenon at 2-C of the 'upper' guibourtinidol moiety. Such an observati may indicate an increased susceptibility to quinone-methide formation at the pyrocatech E-ring under acid-catalysis compared to the phenol B-ring in proguibourtiniding 1, 3, 5 and  $\mathbf{I}$ 

The sparse group of fisetinidol based proguibourtinidins  $1, 9$ , and  $10$ , their natural occurrence hitherto being restricted to the heartwood of  $\ell$ . mopane, is thus complemented by identification of the three additional analogues  $3, 5,$  and  $7$ . Stereo- and electronic factors governing their in vitro formation presumably also control the is vivo process.

### **EXPERIMENTAL**

<sup>1</sup>H NMR spectra were recorded on a Bruker AM-300 spectrometer in CDC1<sub>3</sub> with TMS as internal standard, Accurate mass estimations were obtained with a Kratos MS80 instrument and CD data in methanol on a Jasco J-20 spectropolarimeter. TLC was performed on precoated Merck Plastic sheets (silica gel 60 PF $_{254}$ , 0.25 mm) and the plates sprayed with H2S04-HCHO (4O:l) after development. Preparative plates (PLC), 20x20 cm, Kieaelgel PF254 (1.0 mm) were air-dried and used without prior activation. CC was on Sephadex LB-20 with EtOH as eluant. Methylations were performed with an excess of CH<sub>2</sub>N<sub>2</sub> in MeOH-diethyl ether over 48h at  $-15^{\circ}$ C, while acetylations were in Ac<sub>2</sub>O-pyridine at anoient temperatures. Evaporations were performed under reduced pressure at  $ca$  60°C.

Proguibourtinidins 3, 5, and 7 *from the heariwood of C. mopane* - The extraction (MeOH), fractionation, and derivatization/purification procedures leading to fraction 1.1.1.2 were fully described in Part  $9^{18}$  and those leading to fractions 1.1.2.2 and 2.2.4 in Part  $8^8$ .

Fraction 1.1.1.2 (6.4 mg) afforded the  $(+)$ -guibourtinidol-(4 $\beta$ ,6)-(-)-fisetinidol *penfanelhyl ether diacelote A* as an amorphous white solid (Found: CJgH.+o01I requires M, 684.2571); 'H NMR (Table); CD [8]293 0, [8]280 -0.3X10;, **PI;82 0:**   $\bar{C}_{3.9}H_{4.0}O_{1.1}$  requires M, 684.2571); <sup>1</sup>H NMR (Table); CD [ $\Theta$ ]<sub>293</sub> 0, [ $\Theta$ ]<sub>280</sub> -0.3x10<sup>4</sup>, [ $\Theta$ ]<sub>252</sub> 0, [ $\Theta$ ]<sub>236</sub> 7.5x10<sup>4</sup>, [ $\Theta$ ]<sub>232</sub> 9.2x10<sup>4</sup>, and [ $\Theta$ ]<sub>223</sub> 0.

Fraction 1.1.2.2 (2.1 mg) gave the  $(+)-$ guibourtinidol- $(4\beta,6)-(+)$ -epifisetinidol pea*lamethyl ether diacetate 8 as a white solid (Found: M\*, 684.2560. CsgH<sub>40</sub>0<sub>11</sub> requires M, contract in the solid (Found: M\*, 684.2560. CsgH<sub>40</sub>0<sub>11</sub> requires M,* 684.2571); <sup>1</sup>H NMR data (Table); CD [ $\Theta$ ]<sub>296</sub> 0, [ $\Theta$ ]<sub>287</sub> -0.6x10<sup>4</sup>, [ $\Theta$ ]<sub>282</sub> -0.4x10<sup>4</sup>, [ $\Theta$ ]<sub>269</sub> -1.7x10<sup>4</sup>, [ $\Theta$ ]<sub>253</sub> 0, [ $\Theta$ ]<sub>237</sub> 17.8x10<sup>4</sup>, and [ $\Theta$ ]<sub>2z6</sub> 0.

The aethyl ether fraction 2.2.4 (75.3 q g) was acetylated and separated by PLC in The methyl ether fraction 2.2.4 (75.3 mg) was acetylated and separated by PLC.<br>
a material at Research of 23.8 q and at Richard at Richard at Richard at Plats.  $C_6H_6$ -hexane-Me<sub>2</sub>CO (70:25:5, x4) to give a main band at Rf 0.35 (23.8 mg). De-acetylation<br>with a 1% solution of KOH in methanol for 1h at 45<sup>0</sup>C followed by PLC in C<sub>6</sub>H<sub>6</sub>-Me<sub>2</sub>CO (85:15)

gave two bands at Rf  $0.45$  (3.0 mg) and  $0.31$  (18.9 mg). The latter band was re-subjected to PLC in hexane-Me<sub>2</sub>CO-EtOAc  $(7:2:1, x12)$  to give two bands at Rf 0.51  $(4.2 \text{ mg})$  and 0.44 (7.6 mg). Acetylation of the latter gave the (+)-guibourtinidol-(4a,6)-(+)-epifise**tinidol pcnlanef~y~ ether diaceinie fi** as a white solid (8.0 mg) (Found: M\*, 684.2556. C<sub>39</sub>H<sub>40</sub>O<sub>11</sub> requires M, 684.2571); <sup>1</sup>H NMR data (Table); CD [⊖]<sub>300</sub> 0, [⊖]<sub>259</sub> 1.0x10<sup>4</sup>  $[0.9]_{279}$  0,  $[0.9]_{268}$   $-0.7x10^4$ ,  $[0.9]_{255}$  0,  $[0.9]_{246}$   $2.0x10^4$ ,  $[0.9]_{240}$  0,  $[0.9]_{238}$   $-15.7x10^4$ ,  $[0.9]_{223}$  $-18.8x10<sup>4</sup>$ , and  $[\Theta]_{208}$  0. Acetylation of the Rf 0.45 band (3.0 mg) band gave the known<sup>'</sup>  $(+)$ -guibourtinidol- $(40,8)$ - $(+)$ -catechin hexamethyl ether diacetate. The Rf 0.51 band  $(4.2)$ mg) gave an additional sample of the same compound.

Synthesis of proguibourtinidins 1, 2, 5, and 7. - (-)-Fisetinidol (250 mg) was dissolved in 0.1M HCl (100 ml), a solution of (+)-guibourtacacidin (100 mg) in 0.1M HCl (20 ml) added, and the mixture stirred for 140h at  $45^{\circ}$ C. The mixture was chilled with ice, extracted with EtOAc (5x100 ml), and the combined extract was dried with NazSO4. Evaporation of the solvent afforded a tan powder (235 ag) which was subjected to CC  $(2.7x90$  cm column, flow rate - 1 ml/min, 16.0 ml fractions, first 600 ml of eluant discarded) to give three fractions<sup>d</sup> , 1 (tubes 6-16, 191.6 mg), 2 (32-42, 23.6 mg), and 3  $(46-54, 12.8$  mg). Fraction 1 comprised of a mixture  $(4:1)$  of  $(-)$ -fisetinidol and  $(+)$ epifisetinidol.

Methylation of fraction 2 followed by PLC in  $C_6H_6-Me_2CO$  (9:1, x2) gave two bands at Rf 0.19 (8.6 mg) and 0.15 (4.5 mg). Acetylation of the Rf 0.19 band afforded the  $(+)$ guibourtinidol- $(4\beta, 6)$ - $(-)$ -fisetinidol pentamethyl ether diacetate  $\frac{4}{4}$ . Similar treatment of the Rf 0.15 band gave the  $(+)$ -guibourtinidol- $(4\beta,6)-(+)$ -epifisetinidol pentamethyl ether diacetate  $\underline{8}$ . The physical data of  $\underline{4}$  and  $\underline{8}$  were identical to those of the corresponding derivatives of the natural products.

Fraction 3 (12.8 mg) was methylated and purified by PLC in  $C_6H_6-Me_2CO$  (8:2) to give a main band at Rf 0.39 (8.9 mg). Acetylation and subsequent PLC in C6H6-Me2CO-MeOH  $(97:2:1, x2)$  gave two bands at Rf 0.33  $(3 \text{ mg})$  and 0.29  $(1.6 \text{ mg})$ . The Rf 0.33 band gave the  $(+)$ -guibourtinidol- $(4a,6)-(-)$ -fisetinidol pentamethyl ether diacetate 2, and the Rf 0.29 band the  $(+)$ -guibourtinidol- $(4\alpha, 6)$ - $(+)$ -epifisetinidol pentamethyl ether diacetate  $\underline{6}$ . These compounds were identical to the same derivatives of the natural products by comparison of their <sup>1</sup>H NMR and CD data. Although the <sup>1</sup>H NMR data of 2 was previously published<sup>7</sup>, it is included in the table for comparative purposes.

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### REFERENCES

- 1. Tindale, M.D. ; Roux, D.G. *hyiochenistry, 1974, J.2, 829.*
- 2. Saayman, H.M.; Roux, D.G. *Biochem. J.*, 1965, 96, 36.
- 3. Steynberg, J.P.; Ferreira, D; Roux, D.G. *leirahedron Letters*, 1983, 24, 4147; J. Clea. \$0~. **, Perkin frons. I,** 1987, 1705.
- 1. Pelter, A; Amenechi, P.I.; Warren, R; Harper S.H. J. Chem. Soc. (C), 1969, 2572.<br>5. Dublin: Soc. (c), 1969, 2572. .<br>5. Dublin: Dublin: Paper S.H. Coampan, and annual paper of the Coamrries of the Coamrries of the Coamrr
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- $\alpha = 0.6$ . Ferreira, D.; D.; D.; D.; Roux, D.; D.; D.; D.; D.; D.; D.; D.; Roux, 1985, a, 1985, Ferreira, D.; Du Preez, I.C.; Wijnmaalen, J.C.; Roux, D.G. *Phytochemistry*, 1985, 24,<br>2415.
- 7. Steenkamp, J.A.; Malan, J.C.S.; Roux, D.G.; Ferreira, D. *J. Chem. Soc., Perkin Trans*

dFollowed by a fraction **(ca** 2.0 mg) comprising tri- and tctra-flavanoid analogues which  $\sim$ kollowed by a fraction (

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- a. Malan, J.C.S.; Steenkamp, J.A,; Steynberg, J.P.; Young, D.A.; Brandt, E.V.; Ferreirs D. *J. &en. Sot., Perkin Tram. I,* **1989,** Part **a,** paper 9/01194C.
- 9. Van der Westhuizen, J.H.; Perreira, D.; Roux, D.G. *J. Chem. Soc.*, *Perkin Trans. 1*, **1981,** 1220.
- **10.**  Botha, J.J.; Ferreira, D.; Roux, D.G. J. Chem. Soc., Perkin Trans. 1, 1981, 1235.
- 11, Steenkamp, J.A.; Perreira, D.; Roux, D.G.; Hull, W.E. J. Chem. Soc., Perkin Trans. 1, 1983, 23.
- 12. Botha, J.J.; Young, D.A.; Ferreira, D.; Roux, D.G. J. Chem. Soc., Perkin Trans. 1, 1981, 1213.
- **13.**  Thompson, R.S.; Jacques, D.; Haslam, E.; Tanner, R.J.N. *J. Chea. Sot.,* Perkin frsss. I, 1972, 1387.
- 14. Viviers, P.M.; Botha, J.J.; Ferreira, D.; Roux, D.G.; Saayman, H.M. J. Chem. Soc., Perkin ?rans. 1, 1983, 17.
- 15. Brown, B.R.; Shaw, M.R. *J. Chem. Soc.*, *Perkin Trans. 1*, 1974, 2036.
- 16. Porter L.J.; Wong, R.Y.; Benson, M.; Chan, B.G. *J. Chem. Res.*, 1986 (5), 86; (*K*), 830.
- 17. Steenkamp, **J.A.;** Ralan, J.C.S.; Ferreira, D. *J. thea.* **SOC.,** Petkia **hns I.,** 1988, 2179.
- 18. Malan, J.C.S.; Young, D.A.; Steynberg, J.P.; Ferreira, D, *1. CbeR. Sac., Perkin ?rans. I,* 1989, Part 9, paper 9/018808.